organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

S. Naveen,^a S. A. Khanum,^b M. Mahendra,^a S. Shashikanth,^b M. A. Sridhar^a* and J. Shashidhara Prasad^a

^aDepartment of Studies in Physics, Mansagangotri, University of Mysore, Mysore 570 006, India, and ^bDepartment of Studies in Chemistry, Mansagangotri, University of Mysore, Mysore 570 006, India

Correspondence e-mail: mas@physics.uni-mysore.ac.in

Key indicators

Single-crystal X-ray study T = 295 KMean $\sigma(C-C) = 0.002 \text{ Å}$ R factor = 0.051 wR factor = 0.145 Data-to-parameter ratio = 12.1

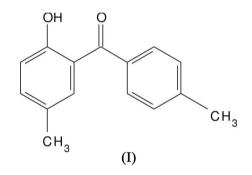
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. (2-Hydroxy-5-methylphenyl)(4-methylphenyl)methanone

In the title compound, $C_{15}H_{14}O_2$, the dihedral angle between the two aromatic rings is 52.75 (7)°.

Received 3 February 2006 Accepted 3 March 2006

Comment

The competence of benzophenones as chemotherapeutic agents, especially as inhibitors of HIV-1 reverse transcriptase RT, cancer and inflammation, is well established and their chemistry has been studied extensively. In addition, methyl-substituted benzophenones exhibit chemotherapeutical activity against fungi. Due to the importance of various substituents on the benzophenone nucleus, the title compound, (I), was synthesized and its crystal structure is reported here.



The molecule of (I) is non-planar (Fig. 1). The dihedral angle between the two aromatic rings is 52.75 (7)°; this compares with a corresponding value of 56.37 (3)° observed for (4-methoxyphenyl)(2-methylphenyl)methanone, (II) (Mahendra *et al.*, 2005). The C4–C5–C7–O16 and O16– C7–C8–C9 torsion angles are 12.9 (2) and 41.1 (2)°, respectively, indicating that the carbonyl group lies nearly in the 2-hydroxy-5-methylphenyl plane but is rather more displaced out of the 4-methylphenyl plane. Bond lengths and angles have normal values and are comparable with those reported for (II). The molecular conformation is stabilized by a strong intramolecular O–H···O hydrogen bond (Table 2). A detailed study of the biological activity of (I) is underway.

Experimental

4-Methylphenyl-4-methylbenzoate (0.022 mol, 5 g) was thoroughly mixed with montmorillonite K-10 clay in the solid state using a vortex mixer and subjected to microwave irradiation at 40% power for 5 min. The completion of the reaction was monitored by thin-layer chromatography and the product was extracted into ether. Finally, the product was isolated and recrystallized from ethanol to afford the title compound (yield: 90%; m.p. 359 K). Analysis calculated: C 79.69, H 6.25, O 14%.

© 2006 International Union of Crystallography All rights reserved

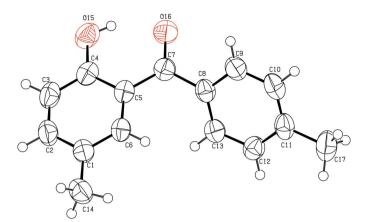


Figure 1

View of (I), with 50% probability displacement ellipsoids.

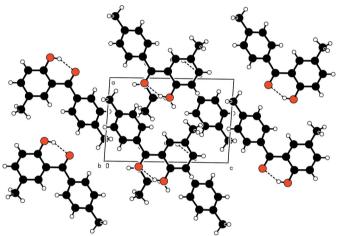


Figure 2

The crystal packing in (I), viewed down the b axis. Dashed lines indicate hydrogen bonds.

Crystal data

$C_{15}H_{14}O_2$	Z = 2
$M_r = 226.26$	$D_x = 1.248 \text{ Mg m}^{-3}$
Triclinic, P1	Mo $K\alpha$ radiation
a = 7.427 (5) Å	Cell parameters from 3046
b = 7.484 (9) Å	reflections
c = 10.940 (13) Å	$\theta = 2.8-25.0^{\circ}$
$\alpha = 88.155 \ (3)^{\circ}$	$\mu = 0.08 \text{ mm}^{-1}$
$\beta = 86.638 \ (7)^{\circ}$	T = 295 (2) K
$\gamma = 82.765 \ (8)^{\circ}$	Block, white
$V = 602.0 (11) \text{ Å}^3$	$0.25 \times 0.2 \times 0.2 \text{ mm}$

Data collection

MacScience DIPLabo 32001 diffractometer ω scans Absorption correction: none 3046 measured reflections 1905 independent reflections

1666 reflections with $I > 2\sigma(I)$
$R_{\rm int} = 0.027$
$\theta_{\rm max} = 25.0^{\circ}$
$h = -8 \rightarrow 8$
$k = -8 \rightarrow 8$
$l = -12 \rightarrow 12$

Refinement

•	
Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0809P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.051$	+ 0.0862P]
$wR(F^2) = 0.145$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.09	$(\Delta/\sigma)_{\rm max} < 0.001$
1905 reflections	$\Delta \rho_{\rm max} = 0.17 \ {\rm e} \ {\rm \AA}^{-3}$
157 parameters	$\Delta \rho_{\rm min} = -0.19 \ {\rm e} \ {\rm \AA}^{-3}$
H-atom parameters constrained	Extinction correction: SHELXL97
	Extinction coefficient: 0.27 (3)

Table 1

Selected	geometric	parameters	(A, °).

O15-C4	1.354 (2)	O16-C7	1.237 (2)
O15-C4-C3	118.06 (14)	O16-C7-C8	118.28 (14)
O15-C4-C5	122.84 (14)	O16-C7-C5	120.61 (14)

Table 2		
Hydrogen-bond geometry	(Å. °).	

$\overline{D-\mathrm{H}\cdots A}$	<i>D</i> -Н	H···A	$D \cdots A$	$D - \mathbf{H} \cdots A$
O15-H15···O16	0.82	1.87	2.580 (4)	145

H atoms were placed at idealized positions and allowed to ride on their parent atoms, with C–H distances in the range 0.92–0.97 Å and O–H = 0.82 Å; $U_{\rm iso}({\rm H})$ values were set equal to $1.2U_{\rm eq}({\rm carrier atom})$.

Data collection: *XPRESS* (MacScience, 2002); cell refinement: *SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *SCALEPACK* and *DENZO* (Otwinowski and Minor, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003) and *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *SHELXL97*.

We thank the DST, Government of India, for financial assistance under the project SP/I2/FOO/93.

References

Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.

MacScience (2002). XPRESS. MacScience Co. Ltd, Yokohama, Japan.

- Mahendra, M., Khanum, S. A., Singh, A., Shashikanth, S., Doreswamy, B. H., Sridhar, M. A. & Shashidhara Prasad, J. (2005). Acta Cryst. E61, o2990– o2991.
- Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.